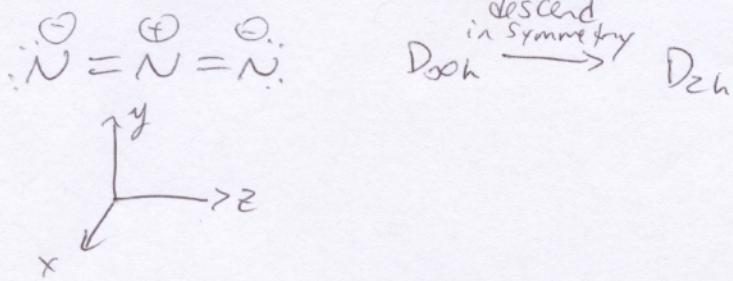
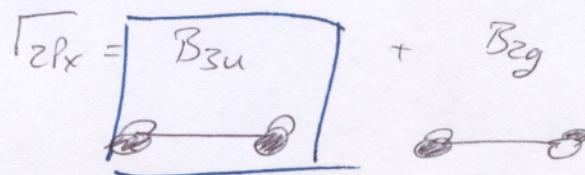
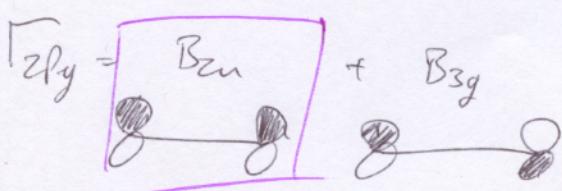
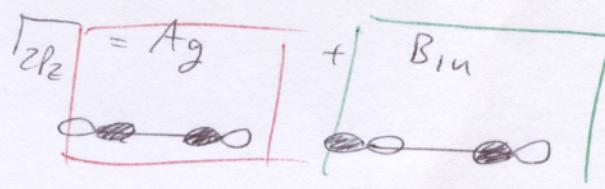
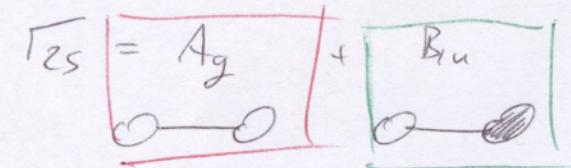


Ch.5 - Exercise 5.7 - Draw an MO Diagram for N_3^-

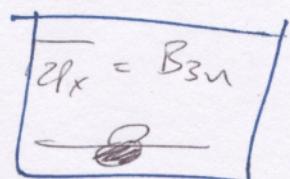
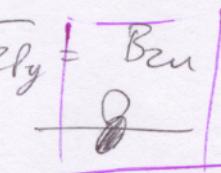
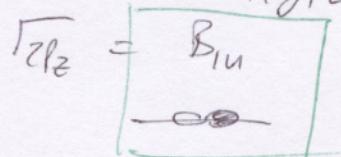
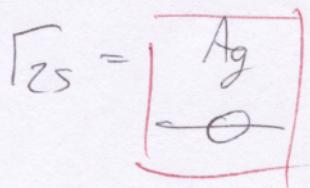


P_{2h}	E	$C_{2v}(z)$	$C_{2v}(y)$	$C_{2v}(x)$	i	σ_{xy}	σ_{xz}	σ_{yz}
Γ_{2s}	2 2	0	0	0	0	2	2	
Γ_{2p_z}	2 2	0	0	0	0	2	2	
Γ_{2p_y}	2 -2	0	0	0	0	-2	2	
Γ_{2p_x}	2 -2	0	0	0	0	2	-2	

Outer Atoms \Rightarrow formed by reducing the reducible representations



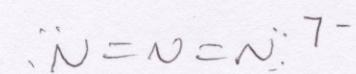
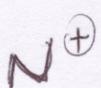
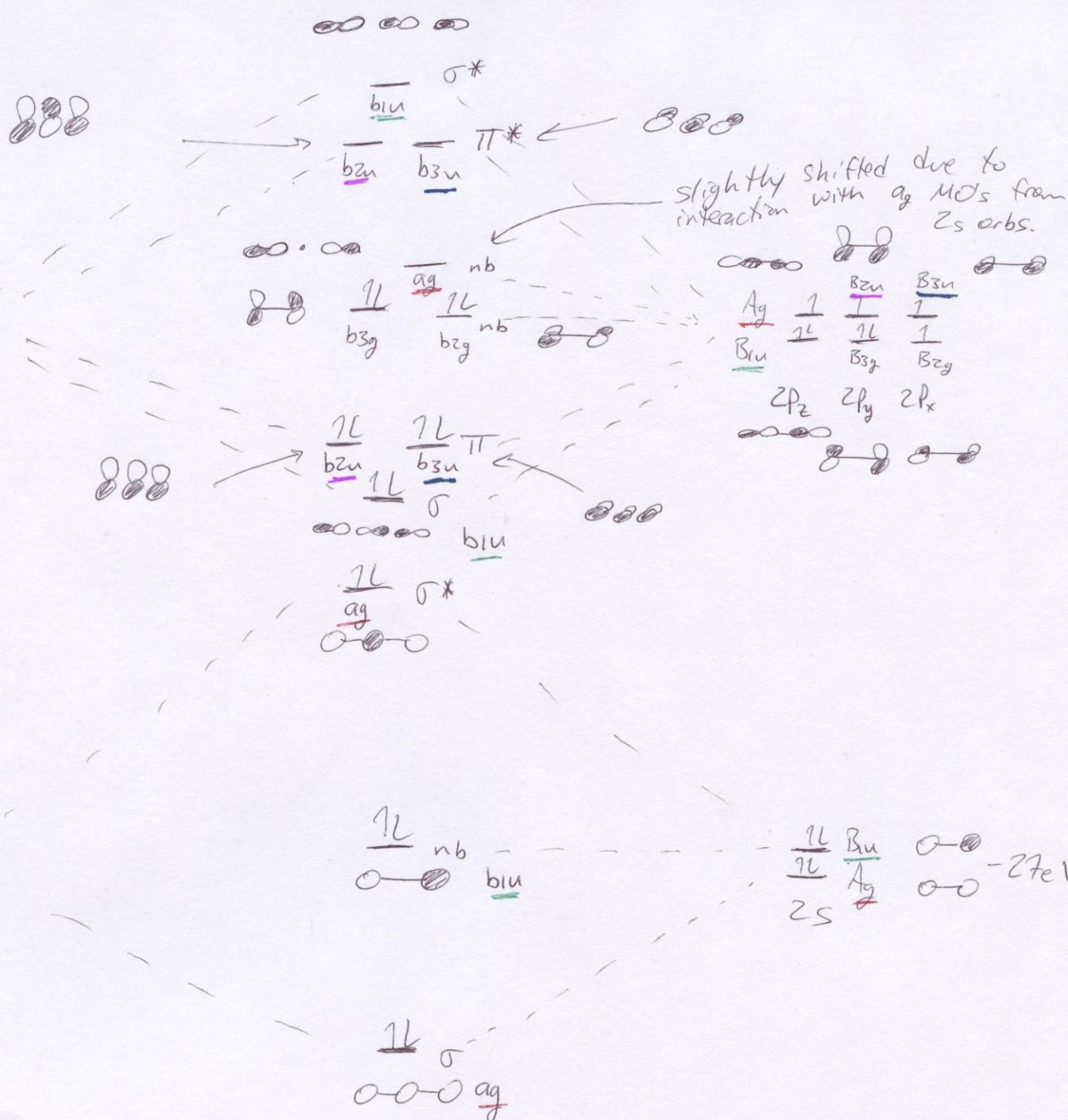
Central Atom \Rightarrow easy to tell symmetries of central atom by looking at the irreducible representations with x, y, z symmetries.



$\hookrightarrow 2p_N = -13eV \Rightarrow 2p$'s will interact with each other

$\hookrightarrow 2s_N = -27eV \Rightarrow 2s$'s will interact with each other.

$\hookrightarrow 2s$'s and $2p$'s will have minimal interaction.



- ↳ 2 σ bonds
- ↳ 2 π bonds
- ↳ 3 lone pairs

} Doesn't fully match our Lewis Dot Structure. One lone pair is likely delocalized. One pair of electrons from our structure is probably delocalized between all 3 N's